

النادي الطبي

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http://www.medclubhu.weebly.com

Outline of biochemistry course

Topic	Likely number of lectures
Enzymes	5
Bioenergetics	1
Electron transport chain	1
Protein metabolism	4
Introduction to metabolism	With first CHO lecture
Carbohydrate metabolism	5
Lipid metabolism	3
Integration of metabolism	1-2

Aim: understand key (simplified) principles (important clinical correlations)

Торіс	Lecture outline
Introduction	 What is biochemistry? Outlines of biochemistry application in medicine

• Biochemistry: science of the chemical basis of life (Gk bios "life")

• It forms a bridge between biology and chemistry

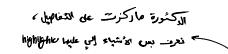
What is biochemistry?

- The cell is the structural unit of living systems
 - \rightarrow biochemistry can also be described as the science of the chemical constituents of living cells & reactions and processes they undergo
- By this definition, biochemistry encompasses large areas of:
 - cell biology
 - molecular biology
 - molecular genetics

Answers "How!" Biochemistry explains to Chemistry (Chemical processes) that occur inside the life (Gk bios "life") If a ganisme (cells)

-> Biochemistry fills the gap

between Chemistry & Biology



Biochemistry applications in medicine

- The biochemistry of the nucleic acids lies at the heart of genetics;
- The use of **genetic** approaches has been critical for elucidating many areas of biochemistry
- **Physiology**, the study of body function, overlaps with biochemistry almost completely
- **Immunology** employs numerous biochemical techniques, and many immunologic approaches have found wide use by biochemists

Biochemistry applications in medicine

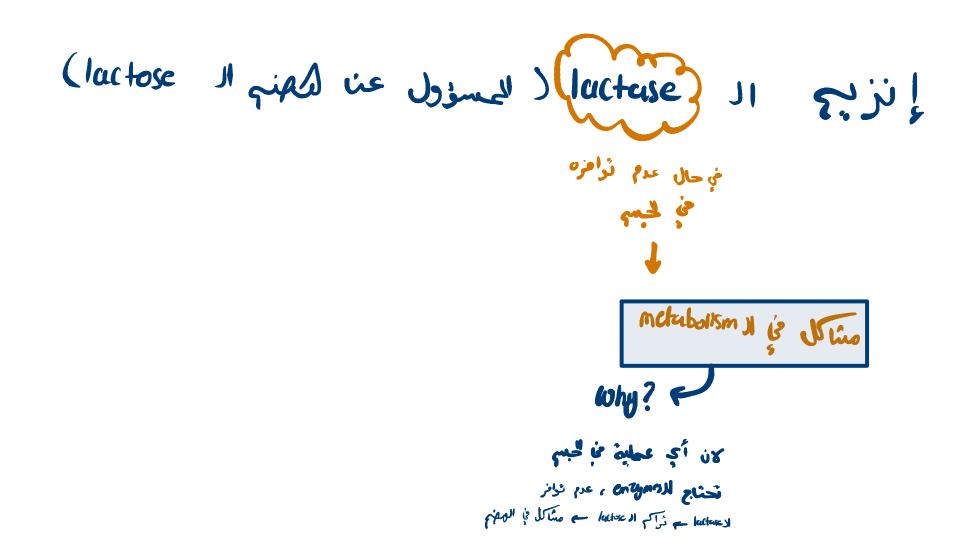
- **Pharmacology** rest on a sound knowledge of biochemistry & physiology;
 - most drugs are metabolized by enzyme-catalyzed reactions
- Poisons act on biochemical reactions or processes; this is **toxicology**
- Biochemical approaches are being used increasingly to study basic aspects of **pathology** (the study of disease), such as inflammation, cell injury, and cancer
- Many workers in **microbiology**, **zoology**, and **botany** employ biochemical approaches almost exclusively

Enzymes I	 1.Understanding enzymes a catalyst 2.The catalytic cycle 3. How enzymes accelerate cellular reactions? 4. The basis of enzyme classifications 5. Exploring the factors affecting the rate of
	enzymic reaction



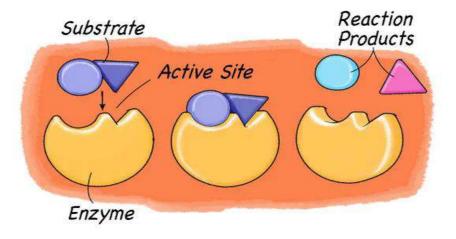
- Definition: Enzymes are specific biocatalysts [mainly proteins in nature] that regulate (accelerate) the rate of biochemical reactions
- Proteins can be hydrolyzed with hydrochloric acid by boiling for a very long time; but inside the body, with the help of enzymes, proteolysis takes place within a short time at body temperature
- <u>Enzyme catalysis is very rapid;</u> usually 1 molecule of an enzyme can act upon about 1000 molecules of the substrate per minute
- ► Lack of enzymes will lead to block in metabolic pathways → inborn errors of metabolism

example :





- The substance upon which an enzyme acts, is called the substrate Substrates are also called <u>reactants</u> because they are the molecules undergoing the reaction
 Substrate = feactant
- The enzyme will convert the substrate into the product or **products**





• Most commonly used enzyme names have the suffix "-ase" attached to the substrate of the reaction (e.g. glucosidase, urease, sucrase)

or

- A description of the action performed (e.g. lactate dehydrogenase) and adenylyl cyclase) • converts include approvate
- Some enzymes retain their original trivial names, which give no hint of the associated enzymatic reaction, e.g. trypsin and pepsin

The basis of enzyme classifications

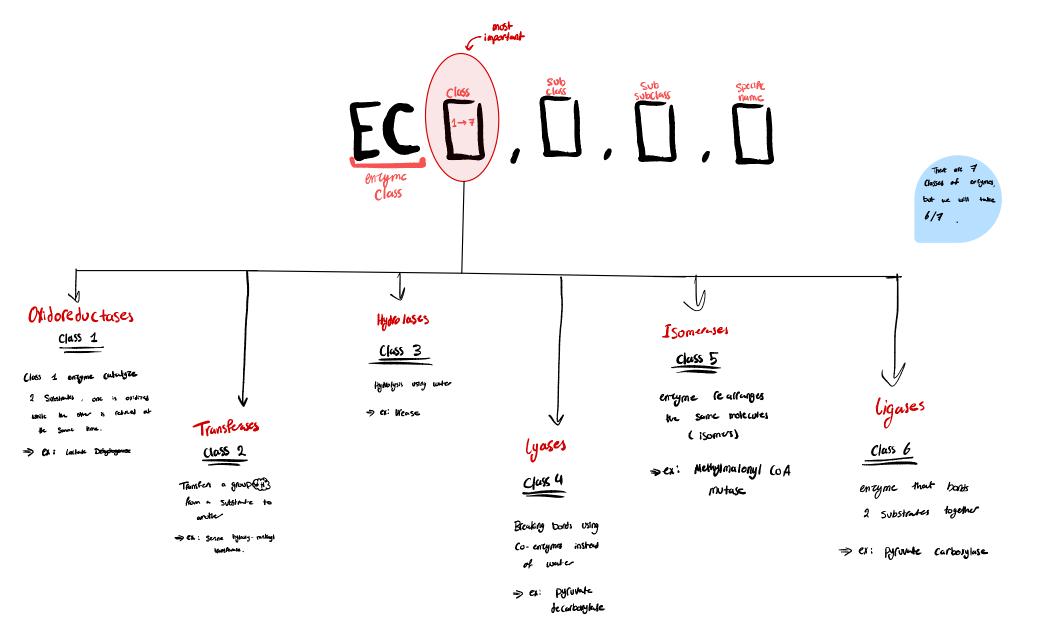
- International Union of Biochemistry and Molecular Biology (IUBMB) developed a system of nomenclature for enzymes
- It is complex and cumbersome; but <u>unambiguous</u>.
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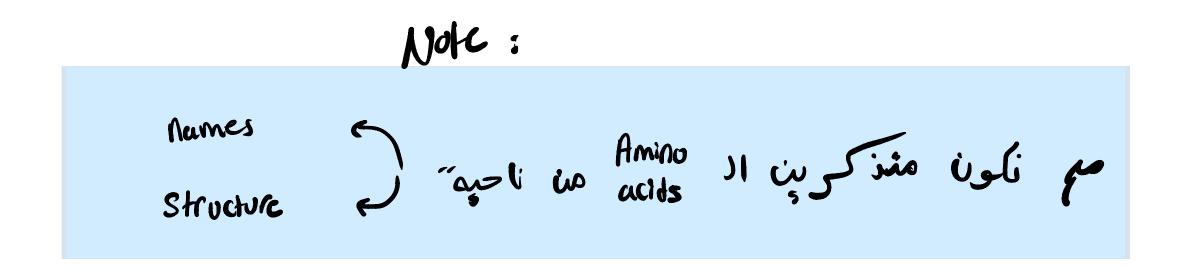
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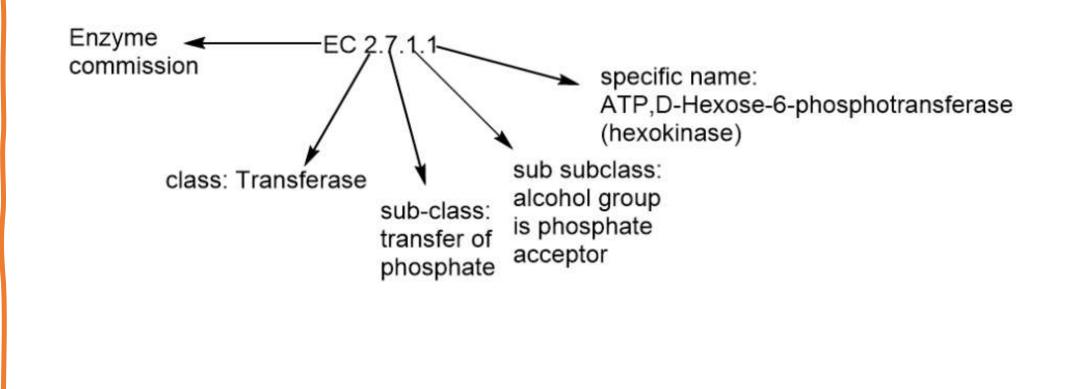
- The name starts with EC (enzyme class) followed by 4 digits:
 - First digit represents the class (6 classes)
 - Second digit stands for the subclass
 - Third digit is the sub-subclass or subgroup
 - Fourth digit gives the number of the particular enzyme in the list

https://www.enzyme-database.org/class.php

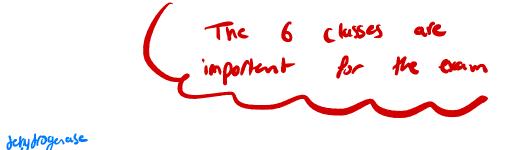
Enzyme nomenclature according to IUBMB



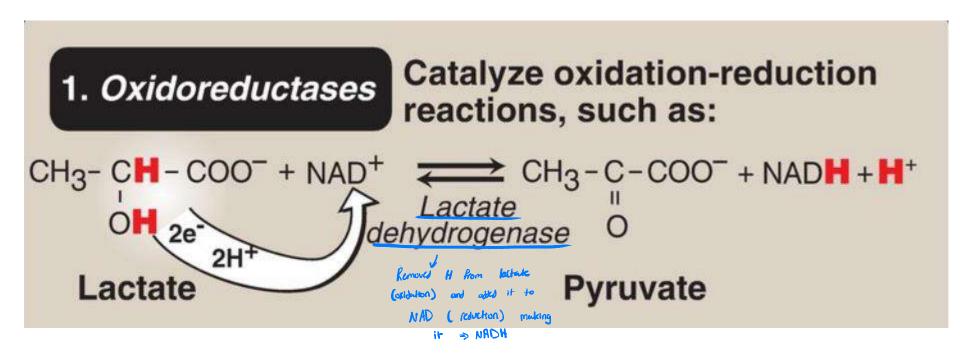




Class 1: Oxidoreductases

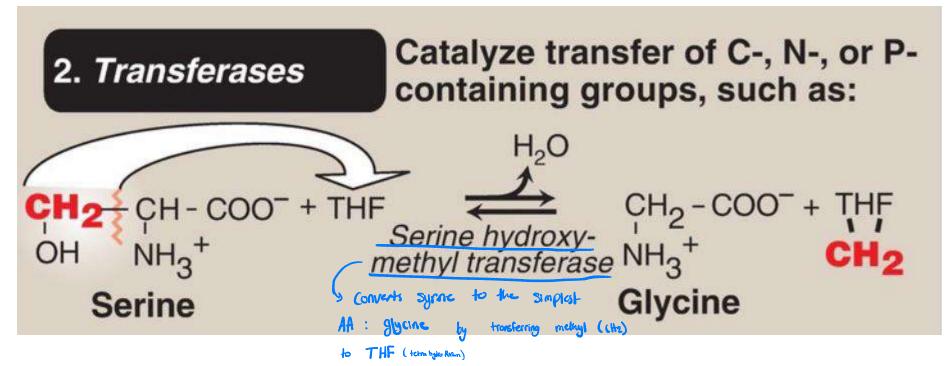


- This group of enzymes will catalyze <u>oxidation of one substrate</u> with simultaneous <u>reduction of another substrate</u> or co-enzyme
 - $AH2 + B \rightarrow A + BH2$



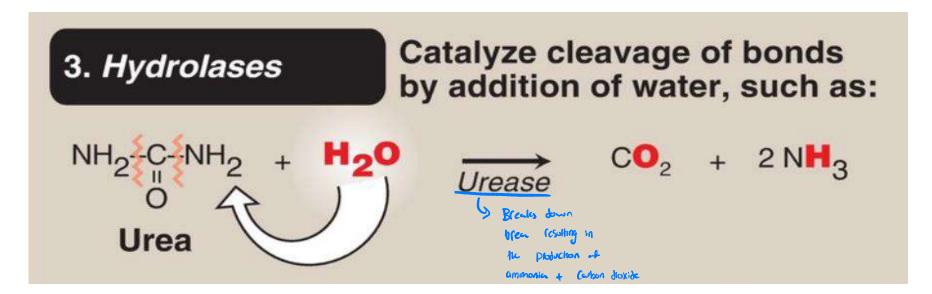
Class 2: Transferases (fransfers a group then one substrate to another)

- This class of enzymes <u>transfers one group</u> (other than hydrogen) from the substrate to another substrate
 - This may be represented as:
 - $A-R + B \rightarrow A + B-R$





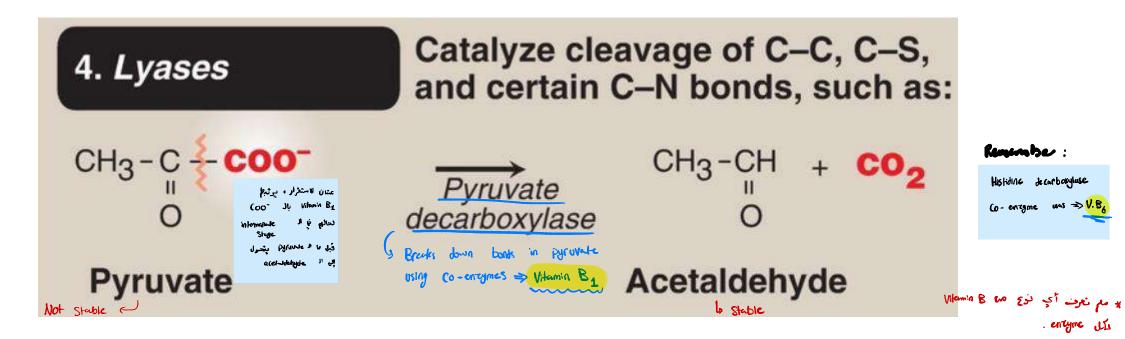
- This class of enzymes can hydrolyze ester, peptide or glycosidic bonds by <u>adding</u> <u>water</u> and then <u>breaking the bond</u>
- All digestive enzymes are hydrolases
- $A-B + H_2O \rightarrow A-OH + B-H$



Class 4: Lyases = Breaking down bonds using methods

- These enzymes can <u>remove groups</u> from substrates or <u>break bonds</u> by mechanisms <u>other than hydrolysis</u>
- ATP \rightarrow cAMP + PPi

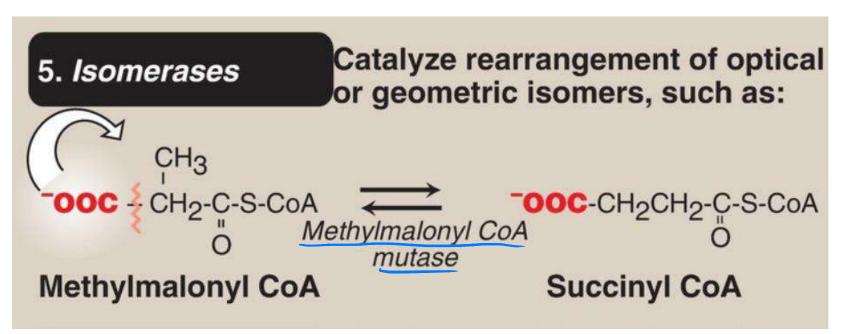
ex : Vitamin B

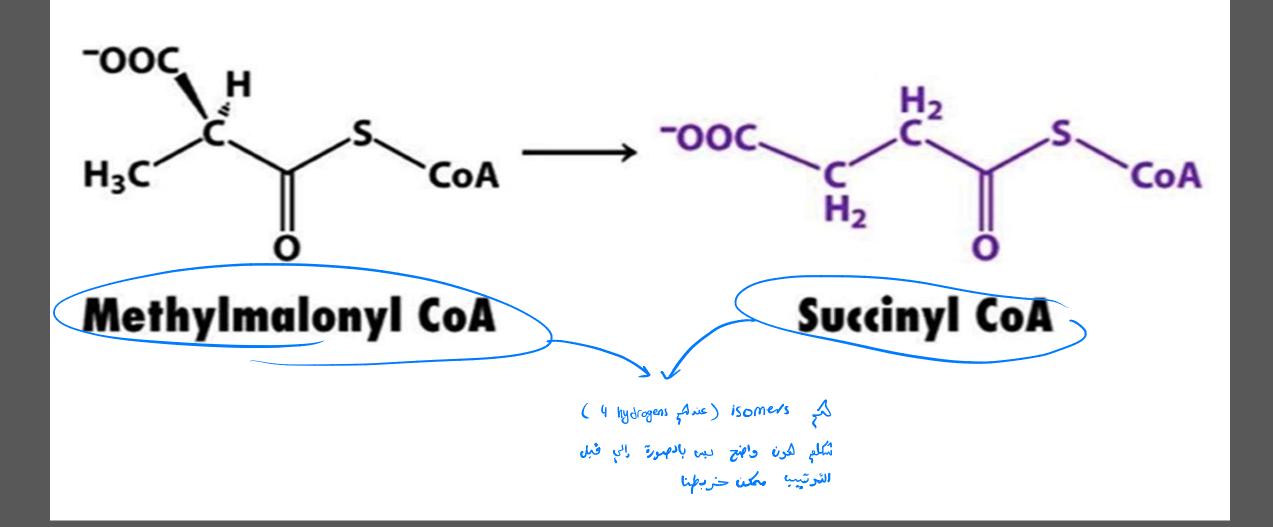


EC 1	[+] Oxidoreductases
EC 2	[+] Transferases
EC 3	[+] Hydrolases
EC 4	[-] Lyases
EC 4.1	[-] Carbon-carbon lyases
EC 4.1.1	[-] Carboxy-lyases
EC 4.1.1.1	pyruvate decarboxylase
EC 4.1.1.2	oxalate decarboxylase
EC 4.1.1.3	oxaloacetate decarboxylase. Now recognized to be two enzymes EC 7.2.4.2 [oxaloacetate decarboxylase (Na ⁺ extruding)] and EC 4.1.1.112 (oxaloacetate decarboxylase).
EC 4.1.1.4	acetoacetate decarboxylase
EC 4.1.1.5	acetolactate decarboxylase
EC 4.1.1.6	cis-aconitate decarboxylase
EC 4.1.1.7	benzoylformate decarboxylase
EC 4.1.1.8	oxalyI-CoA decarboxylase
EC 4.1.1.9	malonyl-CoA decarboxylase
EC 4.1.1.10	
EC 4.1.1.11	
EC 4.1.1.12	
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EC 4.1.1.20	
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EC 4.1.1.30	
EC 4.1.1.31	
LU 4.1.1.JZ	

Class 5: Isomerases = entyme that rearranges the same molecules to produce

- These enzymes can produce isomers of substrates
- Racemases, epimerases, cis-trans isomerases are examples
- $A-B \rightarrow B-A$





Class 6: Ligases = Bonds 2 things together using ATP

- These enzymes <u>link two substrates together</u>, usually with the simultaneous hydrolysis of ATP (Latin, Ligare = to bind)
- A-OH + B-H \rightarrow A-B + H₂O

