

ATHAR BATCH

BIOCHEMISTRY

lecture : 1

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Erythrocytes Metabolism

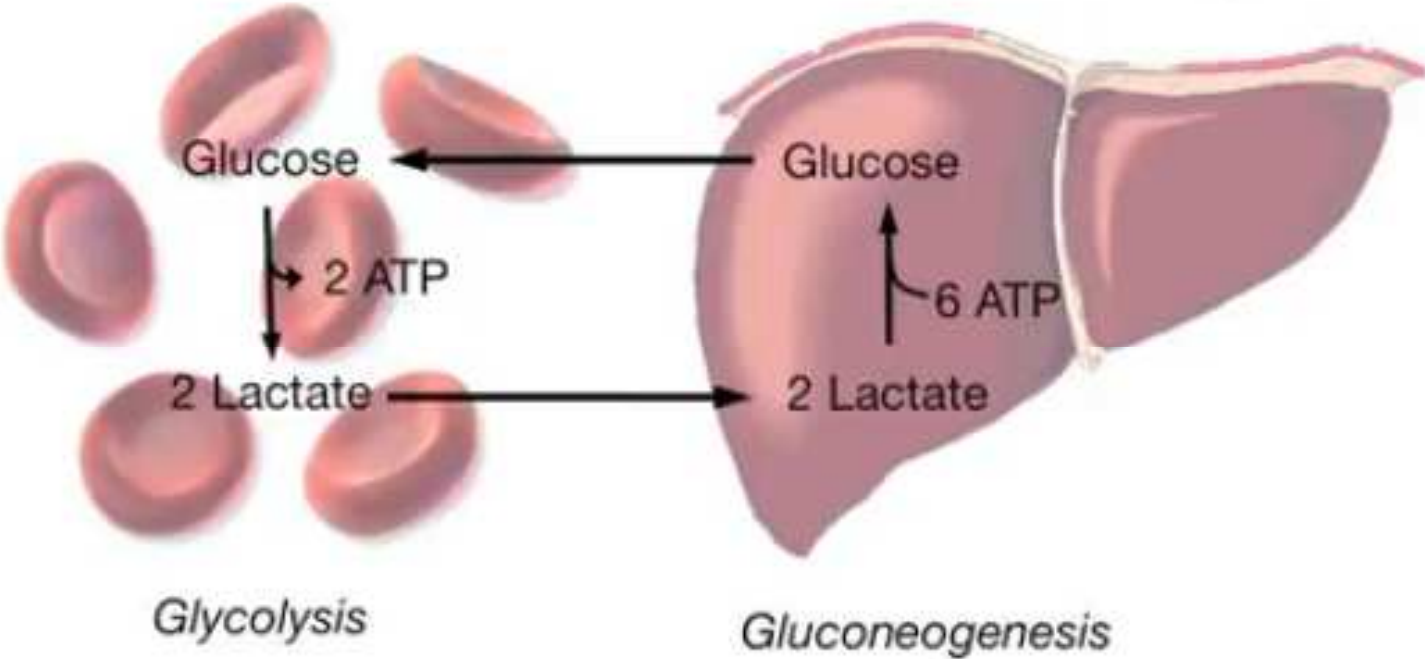
By

Dr. Wasaa Bayoumie El Gazzar

Glycolysis in red blood cells

Pentose phosphate pathway

Cori Cycle



* We all know that the function of RBCs is to transfer O_2 to the Body tissues. if the RBCs contain mitochondria, it will consume the O_2 in glycolysis. that is why it is unfavorable for RBCs to contain Mitochondria.

* the erythrocytes are dependent on glycolysis alone to produce ATP.

* Glycolysis $\begin{cases} \rightarrow \text{Aerobic} \Rightarrow \text{Results in the formation of pyruvate.} \\ \rightarrow \text{Anaerobic} \Rightarrow \text{Results in the formation of Lactic acid} \end{cases}$

* Aerobic glycolysis \rightarrow in the presence of Mitochondria & O_2 .

* Anaerobic glycolysis \rightarrow in the absence of Mitochondria or O_2

* The amount of ATP that results from aerobic g. is larger than that from anaerobic glycolysis.

* accumulation of lactate leads to acidosis, thus affecting RBCs function.

function of RBCs.

* Lactate must be eliminated by gluconeogenesis in the liver.

* Cori cycle: is the metabolic loop that occurs between RBCs & the liver (glycolysis → Formation of Lactate → Transfer of Lactate to the liver → gluconeogenesis → formation of glucose → utilized in glycolysis)

* gluconeogenesis process has a high energy cost. (we need 6 ATP to convert 2 molecules of lactic acid to one molecule of glucose).

Glycolysis

(Embden-meyerhof pathway)

- **Definition:**

- It is oxidation of glucose or glycogen to pyruvic acid (in presence of O_2) or lactic acid (in absence of O_2).

- **Site:**

- It occurs in the cytosol of every cell. Glucose can only give lactic acid in:

- RBCs (no mitochondria).
- Exercising muscles (O_2 lack).

- **Steps:** The steps of glycolysis can be classified into two phases:



PHASE ONE:

In this phase glucose is converted into two molecules of glyceraldehydes-3-phosphate.

PHASE TWO:

In this phase the 2 molecules of glyceraldehydes-3-phosphate are converted into two molecules of pyruvate (aerobic) or lactate (anaerobic).

* Phase one \Rightarrow investment stage

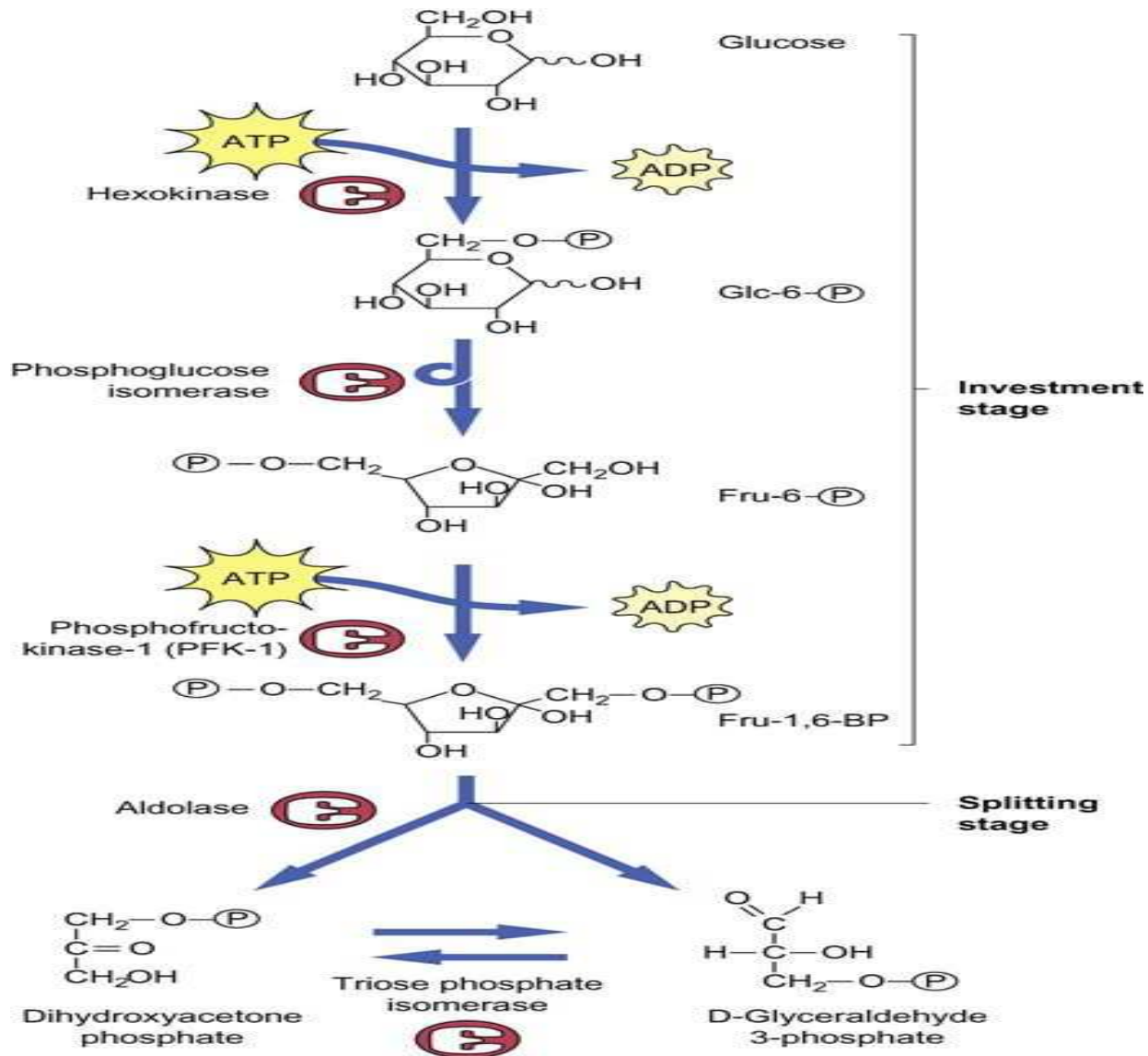
this phase require 2ATP to prime the glycolysis process.

* Phase two \Rightarrow ~~yield~~ yield stage

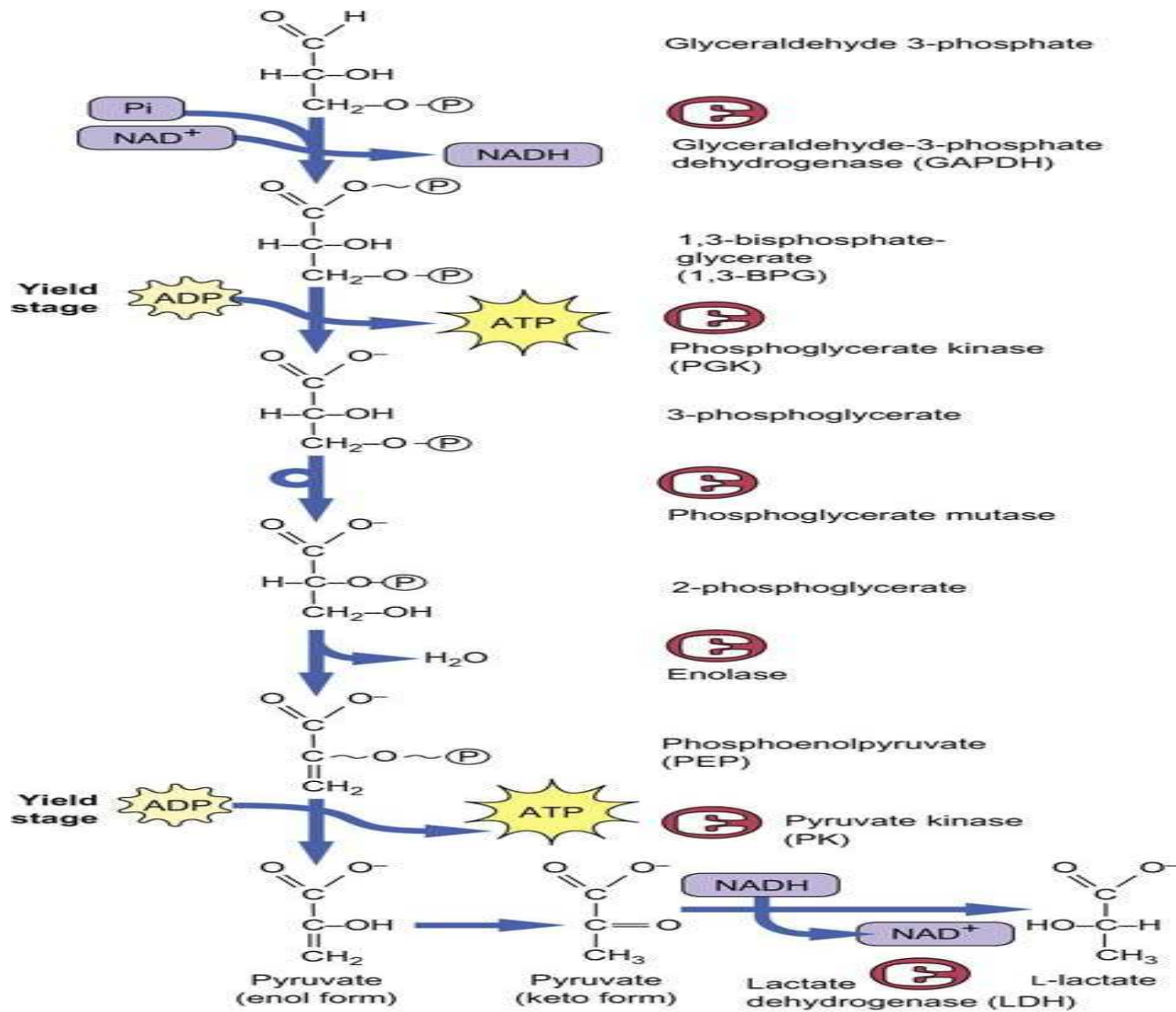
Results in formation of Lactic acid

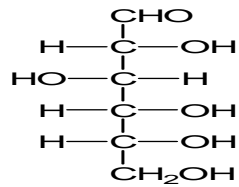
The investment stage of glycolysis

2 ATP are invested to prime the metabolism of glucose by glycolysis

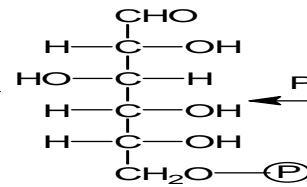
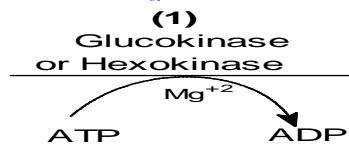


The yield stage of glycolysis:

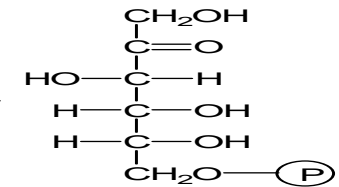
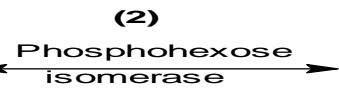




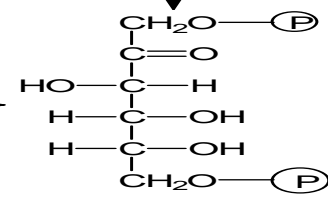
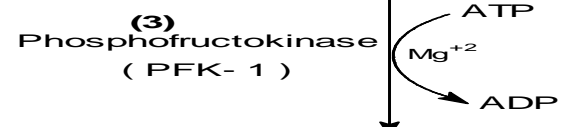
Glucose



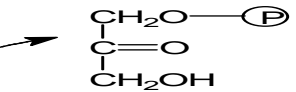
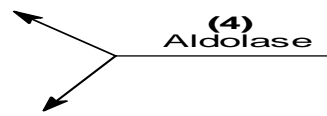
Glucose-6-phosphate



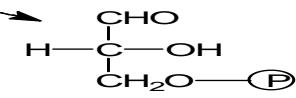
Fructose-6-phosphate



Fructose-1,6-Bisphosphate
 (F-1,6- BP)



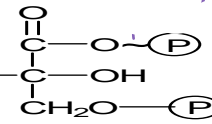
Dihydroxyacetone
 phosphate (DHAP)



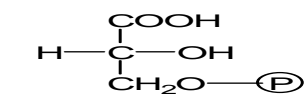
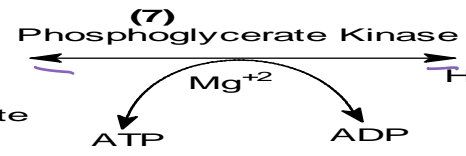
Glyceraldehyde-3-phosphate



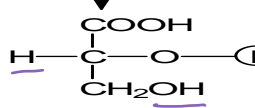
The Embden-Meyerhof
 (Glycolysis) Pathway



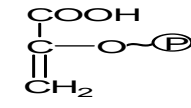
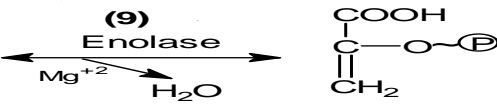
1,3-Bisphosphoglycerate



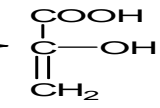
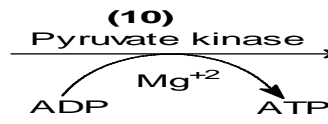
3- Phosphoglycerate



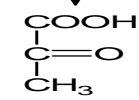
2- Phosphoglycerate



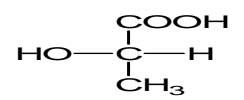
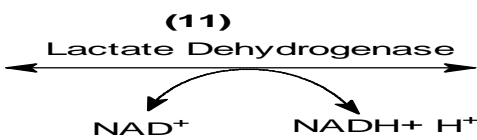
2-phosphoenolpyruvate



Enolpyruvic



Pyruvic acid



Lactic acid

1) Glucose is phosphorylated to form Glucose-6-phosphate by the enzyme Glucokinase or Hexokinase, with utilization of ATP.

* the phosphate group attached to glucose is low energy phosphate.

2) Isomerase enzyme then rearrange the glucose-6-phosphate molecule to get Fructose-6-phosphate.

* the phosphate group is low energy phosphate

3) phosphofructokinase-1 adds a phosphate group on C1 of Fructose-6-phosphate to form Fructose-1,6-Bisphosphate with utilization of ATP

4) Formation of ① Dihydroxyacetone phosphate & ② glyceraldehyde-3-phosphate by Aldolase enzyme.

5) Dihydroxyacetone phosphate (DHAP) is converted to Glyceraldehyde-3-phosphate by isomerase enzyme.

6) Glyceraldehyde-3-phosphate Dehydrogenase \Rightarrow converts Glyceraldehyde-3-phosphate to 1,3-Bisphosphoglycerate by utilization of NAD^+ .

* the carboxylic phosphate ($\text{COOH} + \text{P}$) is a high ^{energy} phosphate.

* NADH when enter the Mitochondria Produce 3 ATP.

7) phosphoglycerate kinase enzyme utilizes the high energy phosphate in 1,3-Bisphosphoglycerate to produce ATP and form

3-phosphoglycerate.

* the phosphate group is low energy phosphate.

8) Mutase enzymes \Rightarrow Rearrangement of 3-phosphoglycerate to form 2-phosphoglycerate.

* Note :

Carboxylic phosphate is a high energy phosphate, while ester Bond phosphate is a low energy phosphate.

- enol phosphate \rightarrow high energy phosphate

9) Enolase enzyme (Needs Mg^{+2}) \Rightarrow forms 2-phosphoenolpyruvate.

* enol phosphate is high energy phosphate

* enol: 2 carbons bound to each other by a double bond, one of the carbons pairing with a hydroxyl group

10) Pyruvate kinase utilizes the high energy phosphate to produce ATP. and forms Enolpyruvic.

11) enolpyruvate is converted spontaneously to pyruvic acid.

12) Lactate dehydrogenase \Rightarrow Converts pyruvic acid to lactic acid

By utilization of $NADH + H^+$ (this reaction is reverse)

Energy gain of glycolysis:

- Energy consumed:

Step (1) by glucokinase: One ATP is lost (spared if we start with glycogen).

Step (3) by phosphofructokinase: One ATP is lost. So, the total lost 2 ATPs

- Energy gained:

Step (6) by glyceraldehyde -3 P dehydrogenase: 2 NADH+H⁺ (6 ATPs) gained only in the presence of O₂.

Step (7) by phosphoglycerokinase: 2 ATPs gained.

Step (10) by pyruvate kinase: 2 ATPs gained. So, the total gains 10 ATPs.

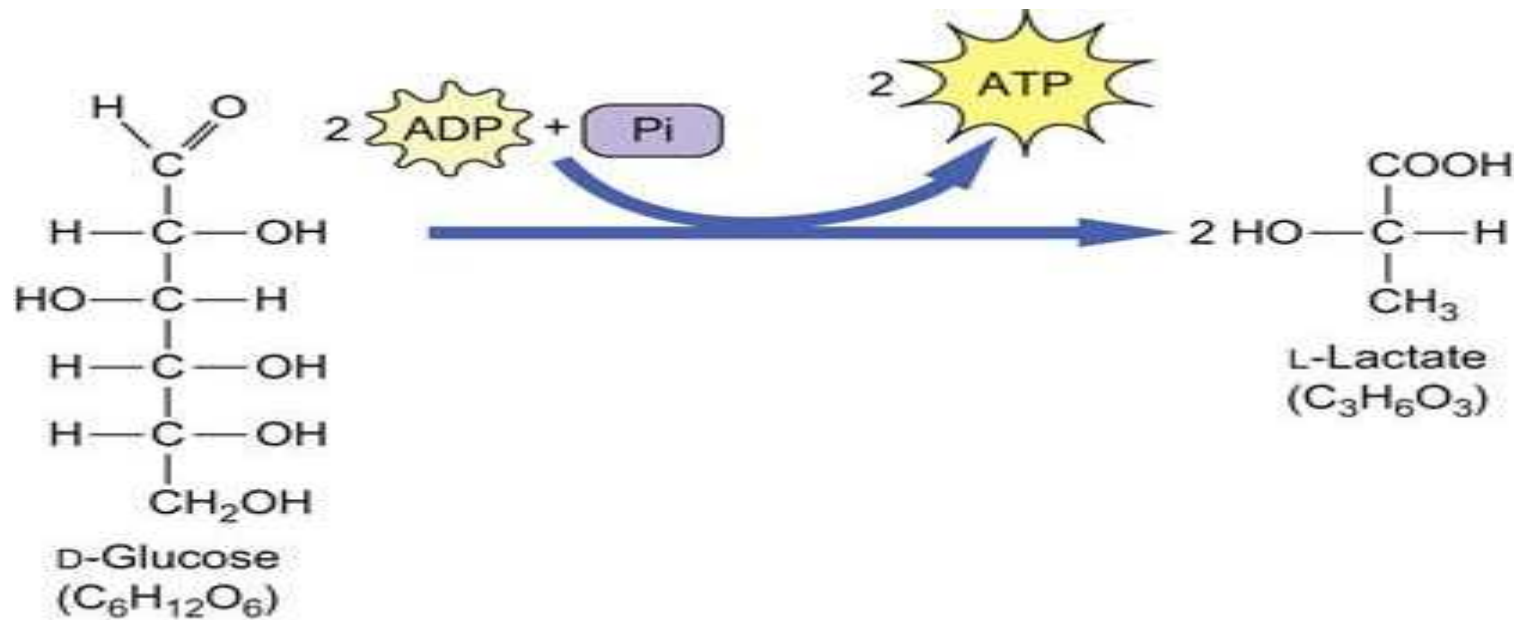
So, Energy gained under anaerobic condition (i.e.) Glucose to 2 molecules of lactic acid is 2 ATPs
and 3 ATPs if we start with glycogen.

Energy gained under aerobic condition (i.e.) Glucose to 2 molecules of pyruvic acid and 2 NADH +H⁺

= 2 ATPs + 6 ATPs (from 2 NADH+H⁺) = **8 ATPs** and 9 ATPs if we start with glycogen.

Glycolysis in red blood cells

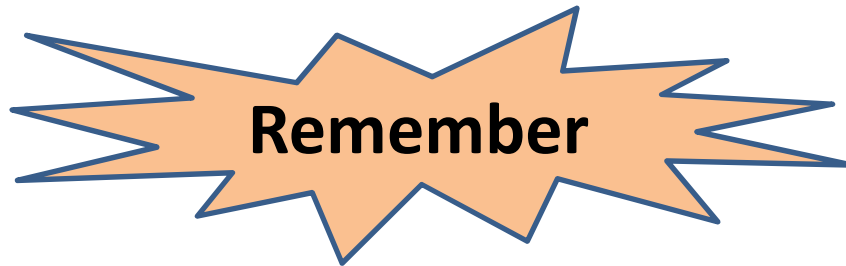
- RBCs have no mitochondria so, glucose oxidation by Glycolysis gives **2 lactic acids** and only **2 ATPs**.





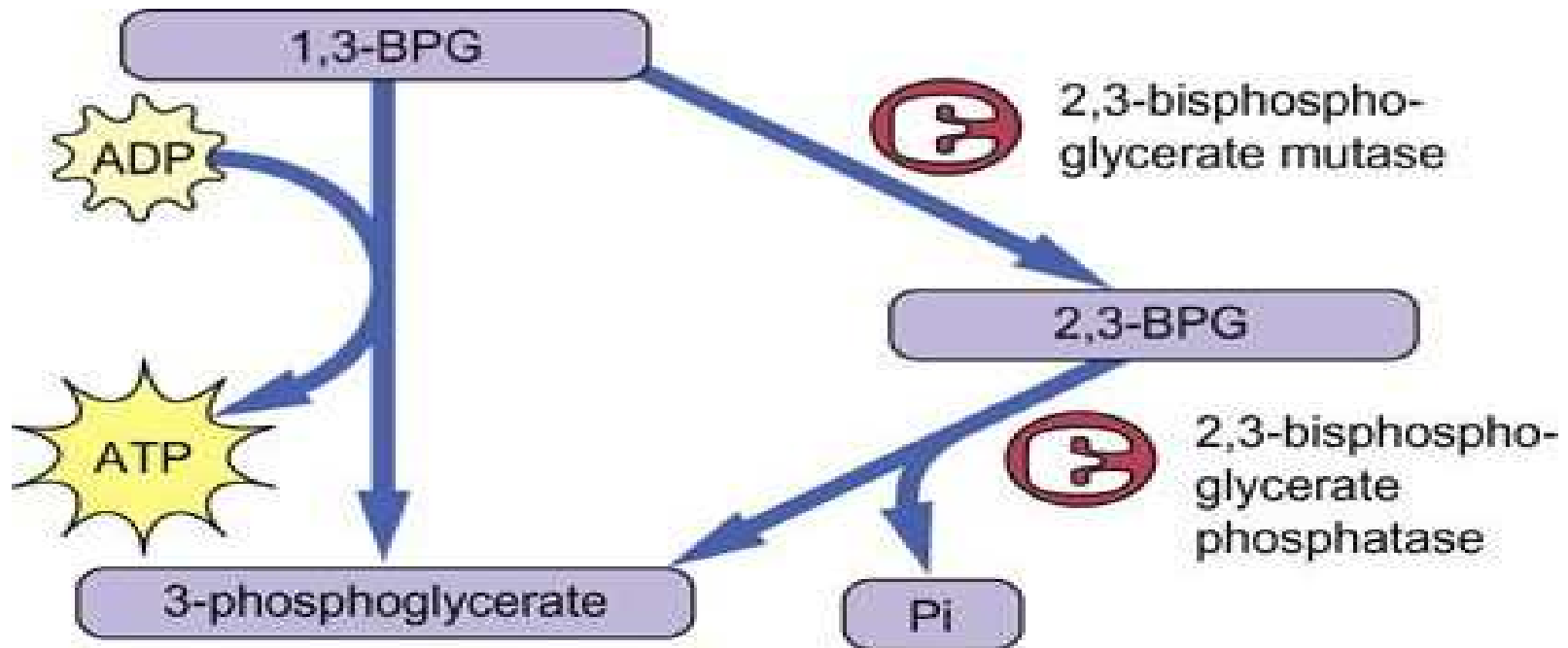
**Sometimes Glycolysis in
RBCs gives NO ATP**

!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!



- The 2,3-bisphosphoglycerate (BPG) molecule carries 5 negative charges and is derived from oxidation of glucose (glycolysis) in red cells.
- It binds to a positively charged pocket in Hb between the 2 β chains (small cavity in the center of the four Hb subunits)
- Binding favors the T- form of Hb, reducing affinity for oxygen and helping delivery of oxygen to tissues.
- BPG increases in red blood cells in cases of chronic anemia and in hypoxia. This helps delivery of oxygen to tissues.

Pathway for biosynthesis and degradation of 2,3-bisphosphoglycerate.



•This pathway discovered by Rapoport-Lubring and called Rapoport-Lubring cycle.

•About 15 to 25% of the glucose utilized in red cells is utilized through BPG shunt.

* Anaerobic glycolysis may not produce any ATP, and this occurs when 2,3-Bisphosphoglycerate is formed.

* 1,3-Bisphosphoglycerate is converted by 2,3-Bisphosphoglycerate mutase enzyme.

- So RBCs need 2, 3 biphosphoglycerate as its increase will decrease the oxygen affinity for hemoglobin to oxygen and helping oxygen delivery to tissues.
- RBCs 1, 3 diphosphoglycerate is changed by mutase to 2, 3 diphosphoglycerate which by phosphatase is changed to 3-phosphoglycerate to continue glycolysis till pyruvic acid.

• Importance of glycolysis in Red cells:

- Energy production: the only pathway that supplies the red cells with ATP.
- Bisphosphoglycerate shunt (BPG shunt).
- Reduction of methemoglobin: glycolysis provides NADH for reduction of met-Hb in red cells by the NADH-cyt.b5-methemoglobin reductase system.

- The ferrous iron of hemoglobin is susceptible to oxidation by superoxide and other oxidizing agents, forming methemoglobin, which cannot transport oxygen.
- Only a very small amount of methemoglobin is present in normal blood, as the red blood cell possesses an effective system (the NADH-cytochrome b5 methemoglobin reductase system) for reducing heme Fe³⁺ back to the Fe²⁺ state.

* When ferrous iron is oxidized, it is converted to ^{ferric} ~~ferric~~ iron.

* Hemoglobin that carries the ferric iron is called methemoglobin

* Methemoglobin cannot transport oxygen because ferric iron cannot carry O_2 .

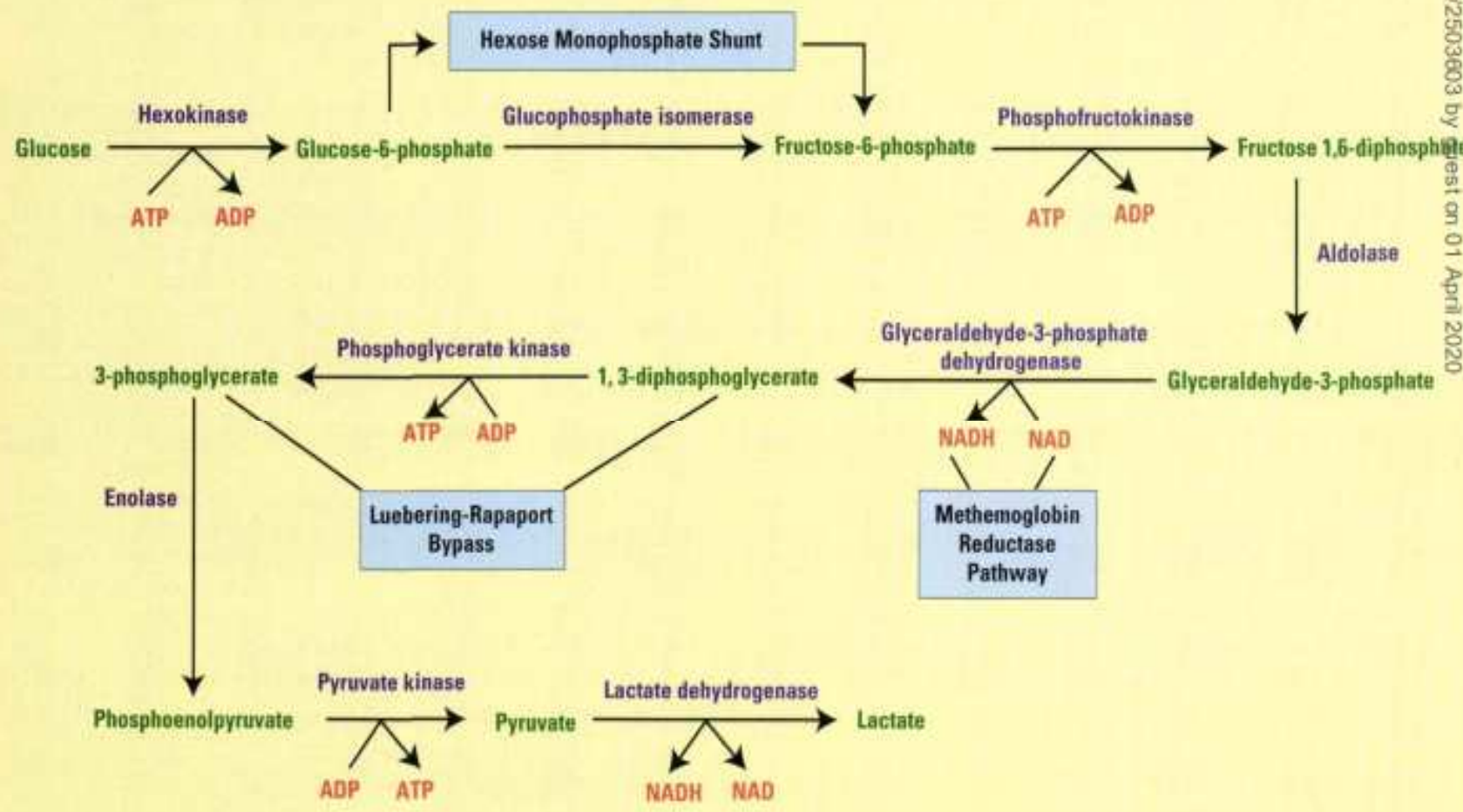
* NADH cytochrome b5 methemoglobin reductase system reduces Fe^{+3} back to Fe^{+2}

- This system consists of NADH (generated by glycolysis), a flavoprotein named cytochrome b5 reductase (also known as methemoglobin reductase), and **cytochrome b₅** (electron transport hemoprotein).
- The Fe³⁺ of methemoglobin is reduced back to the Fe²⁺ state by the action of reduced cytochrome b₅:



- Reduced cytochrome b₅ is then regenerated by the action of cytochrome b₅ reductase (NADH-dependent enzyme):





Hemolytic anemia due to deficiency of glycolytic enzymes:

- Inherited deficiency of glycolytic enzymes produces hemolytic anemia because red cells are dependent on glycolysis for production of ATP.
- About 95% of these patients have deficiency of pyruvate kinase and 4% have deficiency of phosphohexose isomerase.